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college laboratory manual pdf: Practical/Laboratory Manual Chemistry Class XII based on NCERT guidelines by Dr. S. C. Rastogi, Er. Meera Goyal Dr. S. C. Rastogi, Er. Meera Goyal, 2020-06-22 A. Surface Chemistry 1.To prepare colloidal solution (sol) of starch, 2. To prepare a colloidal solution of egg albumin 3.To prepare colloidal solution of gum, 4. To prepare colloidal solution of aluminium hydroxide [Al(OH)3], 5.To prepare colloidal solution of ferric hydroxide [Fe(OH)3], 6.To prepare colloidal solution of arsenious sulphide [As2S3], 7. To purify a freshly prepared sol by dialysis, 8. To compare the effectiveness of different common oils (Castor oil, cotton seed oil, coconut oil, kerosene oil, mustard oil) in forming emulsions. Viva-Voce B. Chemical Kinetics 1. To study the effect of concentration on the rate of reaction between sodium thiosulphate and hydrochloric acid, 2. To study the effect of temperature on the rate of reaction between sodium thiosulphate and hydrochloric acid, 3. To study the rate of reaction of iodide ions with hydrogen

peroxide at different concentrations of iodide ions, 4. To study the rate of reaction between potassium iodate (KIO3) and sodium sulphite (Na2SO3) using starch solution as indicatorl Viva-Voce C. Thermochemistry 1.Determine the enthalpy of dis solution of copper sulphate (CuSO4.5H2O) in water at Room temperature, 2. To determine the enthalpy of neutralization of the reaction between HCl and NaOH, 3. To determine enthalpy change during the interaction between acetone and chloroform Viva-Voce D. Electrochemistry 1.To study the variation of cell potential in Zn|Zn2+||Cu2+|Cu, with change in concentration of electrolytes (CuSO4 or ZnSO4) at room temperature Viva-Voce E.Chromatography 1.To separate the coloured components (pigment) present in the given extract of leaves and flowers by ascending paper chromatography and find their Rf values, 2. To separate the coloured components present in the mixture of red and blue inks by ascending paper chromatography and find their Rf values, 3.To separate Co2+ and Ni2+ ions present in the given mixture by using ascending paper chromatography and determine their Rf values Viva-Voce F. Preparation of Inorganic Compounds 1. Preparation of double salt of ferrous ammonium sulphate (Mohr's salt) from ferrous sulphate and ammonium sulphate, 2. To prepare a pure sample of potash alum (fitkari), 3. Preparation of crystals of potassium ferric oxalate or pottasium trioxlato ferrate (III) Viva-Voce G. Preparation of Organic Compounds 1. Preparation of iodoform from ethyl alcohol or acetone, 2. Preparation of acetanilide in laboratory, 3. Preparation of b-Naphthol aniline dye, 4. To prepare a pure sample of dibenzalacetone, 5. To prepare a pure sample of p-nitro acetanilide Viva-Voce H. Tests for the Functional Groups Present in Organic Compounds Viva-Voce I. Study of Carbohydrates, Fats and Proteins 1.To study simple reactions of carbohydrate, 2. To study simple reactions of fats, 3. To study simple reactions of proteins, 4. To investigate presence of carbohydrates, fats and proteins in food stuffs Viva-Voce J. Volumetric Analysis 1. To prepare 250 ml of M/10 solution of oxalic acid, 2.To prepare 250 ml of M/10 solution of ferrous ammonium sulphate, 3. Prepare M/20 solution of oxalic acid, with its help find out the molarity and strength of the given solution of potassium permanganate, 4. Prepare M/20 solution of Mohr's salt, using this solution determine the molarity and strength of potassium permanganate solution Viva-Voce K. Qualitative Analysis Viva-Voce INVESTIGATORY PROJECTS 1.To study the presence of oxalate ions in guava fruit at different stages of ripening. 2. To study the quantity of caseine present in different samples of milk. 3. Preparation of soyabean milk and its comparison with natural milk with respect to curd formation, effect of temperature etc.4. To study the effect of potassium bisulphite as food preservative at various concentrations. 5. To study the digestion of starch by salivary amylase and the effect of pH and temperature on it. 6. To study and compare the rate of fermentation of the following materials—wheat flour, gram flour, potato juice and carrot juice. 7.To extract essential oils present in saunf (aniseed), ajwain (corum), illaichi (cardomom).8. To detect the presence of adulteration in fat, oil and butter, 9.To investigate the presence of NO2- in brinjal.

college laboratory manual pdf: National Educators' Workshop: Update 1997. Standard Experiments in Engineering Materials, Science, and Technology, 1998 The experiments related to the nature and properties of engineering materials and provided information to assist in teaching about materials in the education community.

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magnet. SECTION: B EXPERIMENTS 1.To find the value of v for different values of u in case of a concave mirror and to find the focal length, 2.To find the focal length of a convex lens by plotting graph between u and v or 1/u and 1/v. 3.To find the focal length of a convex mirror, using a convex lens.4.To find the focal length of a concave lens, using a convex lens. 5. To determine the angle of minimum deviation for a given prism by plotting a graph between the angle of incidence and angle of deviation, 6. To determine refractive index of a glass slab using a travelling microscope, 7.To find the refractive index of a liquid by using a convex lens and a plane mirror, 8.To draw I-V characteristics curve of a p-n function in forward bias and reverse bias, 9.To draw the characteristics curve of a zener diode and to determine its reverse break down voltage, 10.To study the characteristics of a common-emitter n-p-n or p-n-p transistor and to find out the values of current and voltage gains. SECTION: A ACTIVITIES 1.To measure the resistance and impedance of an inductor with or without iron core, 2.To measure resistance voltage (AC/DC), current (AC) and check continuity of given circuit using multimeter, 3. To assemble a household circuit comprising of three bulbs, three (on/off)switches, a fuse and a power source. 4.To assemble the components of a given electrical circuit. 5.To study the variation in potential drop with length of a wire for a steady current, 6.To draw the diagram of a given open circuit comprising atleast a battery, resistor/rheostat, key ammeter and voltmeter. Make the components that are not connected in proper order and correct the circuit and also the circuit diagram. SECTION: B ACTIVITIES 1.To study effect of intensity of light (by varying distance of the source) on an LDR (Light Depending Resistor), 2.To identify a diode, a LED, a transistor, an IC, a resistor and a capacitor from mixed collection of such items, 3. Use a multimeter to: (i) identify the transistor, (ii) distinguish between n-p-n and p-n-p type transistor, (iii) see the unidirectional flow of current in case of a diode and a LED, (iv) Check whether a given electronic components (e.g diode, transistor or IC) is in working order, 4.To observe refraction and lateral deviation of a beam of light incident obliquely on a glass slab, 5.To observe polarisation of light using two polaroids, 6. To observe diffraction of light due to a thin slit, 7.To study the nature and size of the image formed by : (i) convex lens, (ii) concave mirror on a screen by using candle and a screen for different distance of the candle from the lens/mirror, 8.To obtain a lens combination with the specified focal length by using two lenses from the given set of lenses. SUGGESTED INVESTIGATORY PROJECT 1.To Study Verious factors on which the Internal Resistance/EMF of a cell depends, 2.To study the variations in current following in a circuit containing L.D.R. because of variation. (a) In the power of incomdescent lamp used to illum inate the L.D.R. Keeping all the lamps in fixed position (b) In the Distance of a in condescent lamp (of fixed power) used to illum inate the L.D.R. 3. To find the refractive indeces of (a) Water (b) Oil (Transparent) using a plane mirror, an equiconvex lens (made from a glass of known refractive index) and an adjustable object needle, 4. To design an appropriate logic gate combination for a given truth table. 5. To investigate the relation between the ratio of : (i) Output and Input voltage (ii) Number of turms in secondary coils and primary coils of a self designed transformer. 6.To Investigate the dependence of angle of deviation on the angle of incidence, using a hollow prism filled one by with different transparent fluids, 7. To Estimate the charge induced on each one of the two identical styrofoam balls suspended in a vertical plane by making use of coulomob's Law:, 8.To study the factors on which the self inductance of a coil depends by observing the effect of this coil, when put in series with a resistor (bulb) in a circuit fed up by an a.c. source of adjustable frequency, 9.To study the earth's magnetic field using a tangent galvanometer. APPENDIX Some Important Tables of Physical Constants Logarithmic and other Tables

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infectious disease diagnostics. The Manual offers a clear and practical roadmap for topics ranging from selecting appropriate technologies, instruments, and analytic pipelines to navigating complex interpretive challenges and positioning diagnostic testing services for future clinical and population health needs. Beginning with foundational technologies and their clinical applications, this book offers accessible overviews of each method's potential, implications, and emerging roles. Subsequent sections dive meticulously into details of laboratory setup, design, and operations, empowering readers with hands-on insights for routine and advanced testing methods, including advanced sequencing technologies. It also tackles the nuanced challenges of interpreting and reporting results from cutting-edge diagnostics, including those focused on antimicrobial resistance and metagenomics. The final section explores the broader impact of molecular microbiology on value-based care, with discussions on clinical management, laboratory stewardship, and the future of molecular diagnostics in public health. Comprehensive and forward-looking, the Manual of Molecular Microbiology equips readers with both foundational knowledge and practical expertise, making it an indispensable reference for today's clinical laboratory professionals.

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